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A Novel Mechanistic Modelling Approach for Microbial Selection Dynamics: Towards Improved Design and Control of Raceway Reactors for Purple Bacteria – Supplementary Material*

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A novel mechanistic modelling approach for microbial selection dynamics: towards improved design and control of raceway reactors for purple bacteria –
Supplementary material

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Abstract

This document contains supplementary material for the paper “A novel mechanistic modelling approach for microbial selection dynamics: Towards improved design and control of raceway reactors for purple bacteria” by A. Alloul, A. Moradvandi, D. Puyol, R. Molina, G. Gardella, S.E. Vlaeminck, B. De Schutter, E. Abraham, R.E.F. Lindeboom, and D.G. Weissbrodt, *Bioresource Technology*, vol. 390, p. 129844, Dec. 2023.

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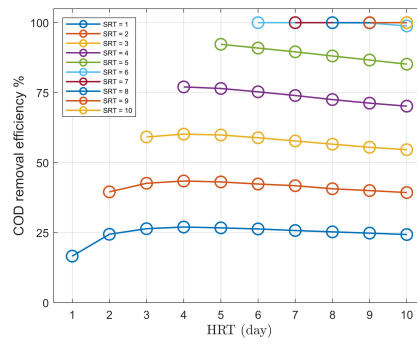
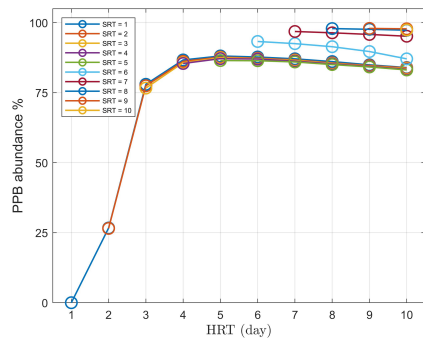


Figure S.1: Effects of HRT and SRT on PPB abundance and COD removal efficiency.

Table S.2: The PBM parameters required for simulation.

| Symbol | Definition | Value | Unit |
|---------------------------|--|---------|-------------------|
| Kinetic parameters | | | |
| $\mu_{m,SS,PB,ph}$ | Maximal specific phototrophic growth rate of PPB on soluble organics | 0.0525 | h^{-1} |
| $\mu_{m,VFA,PB,ph}$ | Maximal specific phototrophic growth rate of PPB on volatile fatty acids | 0.0783 | h^{-1} |
| $\mu_{m,SS,PB,ch}$ | Maximal specific aerobic chemotrophic growth rate of PPB on soluble organic | 0.0500 | h^{-1} |
| $\mu_{m,VFA,PB,ch}$ | Maximal specific aerobic chemoheterotrophic growth rate of PPB on volatile fatty acids | 0.0525 | h^{-1} |
| $\mu_{m,SS,PB,an}$ | Maximal specific anerobic chemoheterotrophic growth rate of PPB on soluble organic | 0.0124 | h^{-1} |
| $b_{m,PB,dec}$ | Specific decay rate of PPB | 0.0113 | h^{-1} |
| $\mu_{m,SS,AHB}$ | Maximal specific growth rate of AHB on soluble organics | 0.0758 | h^{-1} |
| $\mu_{m,VFA,AHB}$ | Maximal specific growth rate of AHB on volatile fatty acids | 0.0758 | h^{-1} |
| $b_{m,AHB,dec}$ | Specific decay rate of AHB | 0.0156 | h^{-1} |
| $\mu_{m,SS,AN}$ | Maximal specific growth rate of AN on soluble organics | 0.0238 | h^{-1} |
| $b_{m,AN,dec}$ | Specific decay rate of AN | 0.00083 | h^{-1} |
| μ_{hyd} | Hydrolysis rate of suspended solids | 0.0035 | h^{-1} |
| $K_{S,E}$ | Light half-saturation constant of PPB | 3 | $W m^{-2}$ |
| $K_{I,E}$ | Light inhibitory constant for chemotrophic growth of PPB | 100 | $W m^{-2}$ |
| $K_{I,O_2,PB}$ | Oxygen inhibitory constant for phototrophic growth of PPB | 0.7 | $mgCODL^{-1}$ |
| $K_{S,O_2,PB}$ | Oxygen half-saturation constant for chemotrophic growth of PPB | 0.05 | $mgCODL^{-1}$ |
| $K_{I,O_2,AHB}$ | Oxygen half-saturation constant for AHB | 0.05 | $mgCODL^{-1}$ |
| $K_{I,O_2,AN}$ | Oxygen inhibitory constant for AN | 0.05 | $mgCODL^{-1}$ |
| $K_{S,SS,ph}$ | Soluble organic half-saturation constant for phototrophic growth of PPB | 5 | $mgCODL^{-1}$ |
| $K_{S,VFA,ph}$ | Volatile fatty acid half-saturation constant for phototrophic growth of PPB | 20 | $mgCODL^{-1}$ |
| $K_{S,SS,ch}$ | Soluble organic half-saturation constant for chemotrophic growth of PPB | 0.4 | $mgCODL^{-1}$ |
| $K_{S,VFA,ch}$ | Volatile fatty acid half-saturation constant for chemotrophic growth of PPB | 0.4 | $mgCODL^{-1}$ |
| $K_{S,SS,an}$ | Soluble organic half-saturation constant for anaerobic chemotrophic growth of PPB | 5 | $mgCODL^{-1}$ |
| $K_{S,SS,AHB}$ | Soluble organic half-saturation constant for AHB | 5 | $mgCODL^{-1}$ |
| $K_{S,VFA,AHB}$ | Volatile fatty acid half-saturation constant for AHB | 5 | $mgCODL^{-1}$ |
| $K_{S,SS,AN}$ | Soluble organic half-saturation constant for AN | 5 | $mgCODL^{-1}$ |
| $K_{S,IN}$ | Inorganic nitrogen half-saturation constant | 0.005 | $mgCODL^{-1}$ |
| $K_{S,IP}$ | Inorganic phosphorus half-saturation constant | 0.001 | $mgCODL^{-1}$ |
| M_S | Metabolic switch | 0.28 | |
| Stoichiometric parameters | | | |
| $Y_{PB,ph}$ | Biomass yield for phototrophic growth of PPB | 1.00 | $mgCODmgCOD^{-1}$ |
| $Y_{PB,ch}$ | Biomass yield for aerobic chemotrophic growth of PPB | 0.52 | $mgCODmgCOD^{-1}$ |
| $Y_{PB,an}$ | Biomass yield for anaerobic chemotrophic growth of PPB | 0.197 | $mgCODmgCOD^{-1}$ |
| Y_{AHB} | Biomass yield for AHB | 0.67 | $mgCODmgCOD^{-1}$ |
| Y_{AN} | Biomass yield for AN | 0.197 | $mgCODmgCOD^{-1}$ |

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| Symbol | Definition | Value | Unit |
|-----------------------------|---|---------------|---|
| $f_{IC,ph,SS}$ | Stoichiometry of inorganic carbon produced for phototrophic growth of PPB on soluble organics | 3.897702e-3 | mgCODmgCOD ⁻¹ |
| $f_{IC,ph,VFA}$ | Stoichiometry of inorganic carbon produced for phototrophic growth of PPB on volatile fatty acids | 3.897702e-3 | mgCODmgCOD ⁻¹ |
| $f_{IC,ch,SS}$ | Stoichiometry of inorganic carbon produced for aerobic chemotrophic growth of PPB on soluble organics | 0.01400 | mgCODmgCOD ⁻¹ |
| $f_{IC,ch,VFA}$ | Stoichiometry of inorganic carbon produced for aerobic chemotrophic growth of PPB on volatile fatty acids | 0.023728 | mgCODmgCOD ⁻¹ |
| $f_{IC,an,SS}$ | Stoichiometry of inorganic carbon produced for anaerobic chemotrophic growth of PPB on soluble organics | -0.02702 | mgCODmgCOD ⁻¹ |
| $f_{IC,AHB,SS}$ | Stoichiometry of inorganic carbon produced for growth of AHB on soluble organics | 0.01400 | mgCODmgCOD ⁻¹ |
| $f_{IC,AHB,VFA}$ | Stoichiometry of inorganic carbon produced for growth of AHB on volatile fatty acids | 0.023728 | mgCODmgCOD ⁻¹ |
| $f_{IC,AN,SS}$ | Stoichiometry of inorganic carbon produced for growth of AN on soluble organics | -0.02702 | mgCODmgCOD ⁻¹ |
| $f_{IC,dec}$ | Inorganic carbon produced from bacterial biomass decay | -1.984127e-04 | mmolHCO ₃ – CmgCOD ⁻¹ |
| $f_{IN,dec}$ | Inorganic nitrogen produced from bacterial biomass decay | 0.058 | mgNH ₃ – NmgCOD ⁻¹ |
| $f_{IP,dec}$ | Inorganic phosphorus produced from bacterial biomass decay | 0.01 | mgPO ₄ – PmgCOD ⁻¹ |
| $f_{SIC,XS}$ | Suspended solids produced from hydrolysis | 1.303971e-06 | mmolHCO ₃ – CmgCOD ⁻¹ |
| $f_{SS,XS}$ | Soluble organics produced from hydrolysis | 1.638241-01 | mgCODmgCOD ⁻¹ |
| $f_{SH2,XS}$ | Hydrogen produced from hydrolysis | 8.442468e-02 | mmolHCO ₃ – CmgCOD ⁻¹ |
| $f_{SIN,XS}$ | Inorganic nitrogen produced from hydrolysis | 1.162246-02 | mgNH ₃ – NmgCOD ⁻¹ |
| $f_{SIP,XS}$ | Inorganic phosphorus produced from hydrolysis | 2.075440e-03 | mgPO ₄ – PmgCOD ⁻¹ |
| $f_{SI,XS}$ | Inert soluble organics produced from hydrolysis | 1.518208e-01 | mgCODmgCOD ⁻¹ |
| $f_{XI,XS}$ | Inert suspended solids produced from hydrolysis | 4.330911e-01 | mgCODmgCOD ⁻¹ |
| $f_{VFA,AN,SS}$ | Fraction of VFA produced during acidogenic fermentation | 0.7728 | mgCODmgCOD ⁻¹ |
| $f_{H2,AN}$ | Fraction of hydrogen produced during acidogenic fermentation | 0.0304 | mgCODmgCOD ⁻¹ |
| $Y_{N,PB}$ | Nitrogen content of PPB | -0.0860 | mgNmgCOD ⁻¹ |
| $Y_{N,AHB}$ | Nitrogen content of AHB | -0.0860 | mgNmgCOD ⁻¹ |
| $Y_{N,AN}$ | Nitrogen content of AN | -0.0860 | mgNmgCOD ⁻¹ |
| $Y_{P,PB}$ | Phosphorus content of PPB | -0.0150 | mgPmgCOD ⁻¹ |
| $Y_{P,AHB}$ | Phosphorus content of AHB | -0.0150 | mgPmgCOD ⁻¹ |
| $Y_{P,AN}$ | Phosphorus content of AN | -0.0150 | mgPmgCOD ⁻¹ |
| Physico-chemical parameters | | | |
| ϵ | Light extinction coefficient | 0.07 | |
| σ | Light absorbance and scattering factor | 1.15 | |
| S_{E0} | Light intensity | 54 | W m ⁻² |
| P_{kaCO2} | Acid-base equilibrium coefficient for inorganic carbon | 6.37 | |
| P_{kaNH4} | Acid-base equilibrium coefficient for inorganic nitrogen | 9.25 | |
| Kla_{O2} | Oxygen gas-liquid transfer coefficient | 1 | h ⁻¹ |
| Kla_{CO2} | Carbon dioxide gas-liquid transfer coefficient | 0.0127 | h ⁻¹ |
| Kla_{NH3} | Ammonia gas-liquid transfer coefficient | 0.533186 | h ⁻¹ |
| Kla_{H2} | Hydrogen gas-liquid transfer coefficient | 1.6 | h ⁻¹ |
| O_2^{sat} | Saturated oxygen | 7.85 | mgO ₂ L ⁻¹ |
| CO_2^{sat} | Saturated carbon dioxide | 0.0127 | mmolHCO ₃ ⁻ L ⁻¹ |
| NH_3^{sat} | Saturated ammonia | 0.533186 | mgNL ⁻¹ |

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| Symbol | Definition | Value | Unit |
|------------------|--|--------------------------------------|--------------------------------------|
| H_2^{sat} | Saturated hydrogen | 1.6 | mgH ₂ L ⁻¹ |
| $f_{O_2, PB}$ | Oxygen uptake chemoheterotrophy of PPE | $\frac{-(1-Y_{PB, ch})}{Y_{PB, ch}}$ | mgO ₂ mgCOD ⁻¹ |
| $f_{O_2, AHB}$ | Oxygen uptake chemoheterotrophy of AHB | $\frac{-(1-Y_{AHB})}{Y_{AHB}}$ | mgO ₂ mgCOD ⁻¹ |
| Reactor geometry | | | |
| V | Volume | 100 | L |
| A | Area | 0.5 | m |
| $f_{H/S}$ | Fraction of removed particles | $\frac{HRT}{SRT}$ | |

Table S.3: Different versions of mechanistic models simulating the utilization of PPB in WWT.

| Name | PAnM | PBM | ePAnM |
|-----------------------------|--|---|---|
| Conditions | Anaerobic | Semi-aerobic | Aerobic |
| Processes | 6 | 16 | 30 |
| <i>PPB-based</i> | 5 | 6 | 11 |
| <i>Other biomass</i> | - | 5 | 19 |
| <i>Hydrolysis</i> | 1 | 1 | 1 |
| <i>Physical processes</i> | - | 4 | 2 |
| State variables | 10 | 15 | 21 |
| <i>PPB-based</i> | 1 | 3 | 1 |
| Substrate | Acetate Other organics | VFAs Other organics | Acetate Other VFAs Other organics |
| Biomass | PPB | Photoheterotrophic PPB (An)aerobic chemoheterotrophic PPB Aerobic heterotrophic Acidogenic | PBB Aerobic heterotrophic Acidogenic Acetogenic Aerobic predators Heterotrophic sulphate reducing bacteria Autotrophic sulphate reducing bacteria Microalgae biomass |
| PPB metabolisms | Photoheterotrophic Photoautotrophic Anaerobic chemoheterotrophic | Photoheterotrophic (An)aerobic chemoheterotrophic | Photoheterotrophic Photoautotrophic (An)aerobic chemoheterotrophic Fermentative |
| Inhibition functions | Competitive inhibitions Inorganic nitrogen Inorganic phosphor | Competitive inhibitions Inorganic nitrogen Inorganic phosphor Photoheterotrophic light Chemoheterotrophic light Photoheterotrophic oxygen Chemoheterotrophic oxygen | Inorganic nitrogen Inorganic phosphor Free amonia Light Oxygen pH Temperature |
| Physical processes | - | Carbon dioxide dissolution/stripping Oxygen dissolution/stripping Stripping of hydrogen Stripping of ammonium | Carbon dioxide loses Oxygen supply |
| Operational mode | Continuous | SBR/Constant | Continuous |
| pH | Dynamic | Constant | Dynamic |
| Temperature | Constant | Constant | Cardial temperature model |
| Light | Constant | Lambert-Beer's law | Lambert-Beer's law |